

# BREAKING THROUGH THE SODIUM ION ENERGY STORAGE BARRIER



How does  $\text{Na}^+$  desolvation and diffusion barrier affect sodium ion storage performance? It's revealed that  $\text{Na}^+$  desolvation and diffusion barrier at the electrode surface and interface play a predominant role on the sodium-ion storage performance. The  $\text{Na}^+$  desolvation barrier in ether electrolytes is less than one third of that in ester electrolytes, leading to enhanced kinetics and remarkably improved ICE.



How do ionic anchoring separators improve the performance of sodium-ion batteries? Enhancing Robustness and Charge Transfer Kinetics of Sodium-Ion Batteries through Introduction of Anionic Anchoring Separators Ionic transport critically dictates the performance of the batteries.



Can ether electrolytes break the innate limitation of sodium ion storage? The innate limitation of sodium-ion storage for nanostructured carbon anode can be broken by neat ether electrolytes. The strong adsorption and decomposition of electrolytes on graphene planes is remarkably reduced in ether solvents due to the small  $\text{Na}^+$  desolvation barrier and decreased Gibbs free energies of adsorption. 1. Introduction



What is  $\text{Na}^+$  desolvation barrier in ether electrolytes?  $\text{Na}^+$  desolvation barrier in ether electrolytes is less than one-third of that in ester electrolytes. The larger surface area of electrode material is, the better performance it delivers in neat electrolytes. Gibbs absorption free energies is a regulation parameter for tailoring electrolytes with material.



Can hard carbon be used for sodium ion batteries? Please reconnect Bridging Microstructure and Sodium-Ion Storage Mechanism in Hard Carbon for Sodium Ion Batteries Hard carbon (HC) has emerged as a strong anode candidate for sodium-ion batteries due to its high theoretical capacity and cost-effectiveness.

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Is PNC a good electrolyte for sodium ion storage? By contrast, PNC in ether electrolytes exhibits remarkably excellent rate capability in ether electrolytes and even at a very high rate of 20 A g<sup>-1</sup>, large capacities of 331.3 mA h g<sup>-1</sup> in DME and 296 mA h g<sup>-1</sup> in DEGDME were retained, demonstrating the fast kinetics of sodium-ion storage.



Na<sub>4</sub>FeV(PO<sub>4</sub>)<sub>3</sub> (NFV) is a Na-super-ionic conductor (NASICON)-structured cathode material for sodium-ion batteries (SIBs). Nonetheless, how to stabilize the V<sup>4+</sup>/V<sup>5+</sup> ???



This study successfully correlates structural attributes with electrochemical performance, shedding light on what makes HC effective for sodium-ion storage. It is found that HC featuring larger interlayer spacing and ???



Results indicate that this strategy effectively anchors free anions and increases the proportion of solvent-separated ion pairs in the bulk, reduces the cation transfer energy barrier ???



Energy storage technology has rapidly developed during the past two decades [1, 2]. With the merits of high energy density and long cycle life, lithium-ion batteries have become ???

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Fig. 2. CV curves sweep rate of 0.1-50 mV s<sup>-1</sup> for PNC electrodes in (a) DME and (b) EC/DMC. The translation of logarithmic format for PNC electrodes in (d) DME and (e) EC/DMC. (c) GITT potential profiles of the two electrodes for sodiation ???



„Breaking the limitation of sodium-ion storage for nanostructured carbon ???



Rechargeable batteries, as the large-scale energy storage systems (ESSs), are essential to ensure a continuous and stable output of renewable and sustainable energy [1]. ???



Pyrite (FeS<sub>2</sub>) is a functional material of great importance for lithium/sodium ion batteries (LIBs/SIBs), but its sluggish dynamics greatly hinder its high performance. Here, we demonstrate an effective strategy of regulating the ???



Breaking the barriers: The tunable c-a interface enhances electron transport, ion diffusion, and energy storage capability. By introducing Vo through NaBH<sub>4</sub> reduction, Fe<sub>3</sub> ???

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The fast transfer kinetics was further confirmed to be originated from the reduction of energy barriers for both de-solvation process and Li + diffusion through SEI, which is ???



The surging demand for lithium-ion batteries (LIBs) has gradually revealed issues such as the limited reserves and high costs of lithium resources, which restrict the widespread ???



Boosting the lithium-ion and sodium-ion storage performances of pyrite by regulating the energy barrier of ion Pyrite ( $\text{FeS}_2$ ) is a functional material of great importance for lithium/sodium ion ???



Solid electrolytes may overcome key technological hurdles associated with the narrow electrochemical and thermal stability of conventional lithium (Li)-ion and sodium (Na)-ion batteries. However, many solid ???



Rechargeable batteries, as the large-scale energy storage systems (ESSs), are essential to ensure a continuous and stable output of renewable and sustainable energy [1].As ???