

# MIRABILITE AS PHASE CHANGE ENERGY STORAGE MATERIAL



Is mirabilite a good energy storage material? Volume 16, article number 220619, ( 2022 ) As a kind of essential hydrated salt phase change energy storage materials, mirabilite with high energy storage density and mild phase-transition temperature has excellent application potential in the problems of solar time and space mismatch.



Are phase change materials suitable for thermal energy storage? Phase change materials (PCMs) having a large latent heat during solid-liquid phase transition are promising for thermal energy storage applications. However, the relatively low thermal conductivity of the majority of promising PCMs ( $<10 \text{ W/(m} \cdot \text{K)}$ ) limits the power density and overall storage efficiency.



What is the latent heat of crystallization of CGCA-supported mirabilite phase change materials? The latent heats of melting and crystallization of CGCA-supported mirabilite phase change materials (CGCA-PCMs) are 157.1 and 114.8 J/g, respectively. Furthermore, after 1500 solid-liquid cycles, there is no leakage, and the retention rate of crystallization latent heat is 45.32%, exhibiting remarkable thermal cycling stability.



Can biobased phase change materials revolutionise thermal energy storage? Low, medium-low, medium, and high temperature applications. An upcoming focus should be life cycle analyses of biobased phase change materials. Harnessing the potential of phase change materials can revolutionise thermal energy storage, addressing the discrepancy between energy generation and consumption.



What are phase change materials? Phase change materials are renowned for their ability to absorb and release substantial heat during phase transformations and have proven invaluable in compact thermal energy storage technologies and thermal management applications.

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How do hydrophilic aerogels support mirabilite phase change materials (CGCA-PCMS)? The hydrophilic aerogels supported MPCMs decrease mirabilite leaking and reduce supercooling to around  $0.7 \pm 1$  °C. The latent heats of melting and crystallization of CGCA-supported mirabilite phase change materials (CGCA-PCMs) are 157.1 and 114.8 J/g  $\pm 1$ , respectively.



As a kind of essential hydrated salt phase change energy storage materials, mirabilite ???



There are two technical problems in using mirabilite alone as phase change materials. One is the super-cooling phenomenon, and the other is the phase separation. This ???



As a kind of essential hydrated salt phase change energy storage materials, mirabilite with high energy storage density and mild phase-transition temperature has excellent application potential in



A technology of solid-liquid and composite phase change of Glauber's Nitrate, which is applied in the direction of heat exchange materials, chemical instruments and methods, etc., can solve ???

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As a new type of energy storage material, phase change material absorbs heat energy as latent heat through its phase change in both solid and liquid states. Additionally, the corrosion of phase change material has been investigated ???



Additionally, the crystalline mirabilite composite phase change material does not exhibit pronounced delamination. 3) The introduction of carbon nanoparticles results in an ???



Phase change materials (PCMs), capable of reversibly storing and releasing tremendous thermal energy during nearly isothermal and isometric phase state transition, have received extensive attention in the fields of energy storage and conversion ???



As a kind of essential hydrated salt phase change energy storage materials, mirabilite with high energy storage density and mild phase-transition temperature has excellent application in various fields ???



Herein, poly (vinyl alcohol) (PVA) aerogels with hierarchically porous structure are fabricated and introduced into phase change matrices to obtain novel leakage-proof composite phase change materials ???

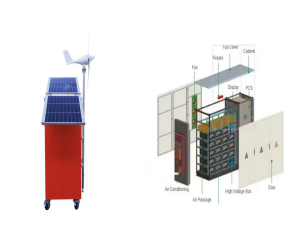
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? 1/4 ? As a kind of essential hydrated salt phase change energy storage materials, mirabilite with high energy storage density and mild phase-transition temperature has excellent application ???



Given its issues with phase stratification and supercooling degree, mirabilite phase-change energy storage material, a type of inorganic hydrated salt with a high latent heat value ???



Sodium sulfate decahydrate ( $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$ ), also known as mirabilite or Glauber's salt, has been the most investigated salt hydrate for use in latent thermal energy ???



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Sodium sulfate decahydrate ( $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$ ), also known as mirabilite or Glauber's salt, has been the most investigated salt hydrate for use in latent thermal energy storage systems since the earliest works of Telkes [2], mostly ???